

**ESSAY. Write your answer in the space provided or on a separate sheet of paper.**

- 10) By using the appropriate molecular formulas, write a balanced equation which describes the complete combustion of nonane.
- 11) View a butane molecule along the C<sub>2</sub>–C<sub>3</sub> bond and provide a Newman projection of the lowest energy conformer.
- 12) Use a sawhorse structure to depict the eclipsed conformer of ethane.
- 13) Provide a representation of the gauche conformer of butane.
- 14) Describe the source of angle strain and torsional strain present in cyclopropane.
- 15) Draw the most stable conformation of *trans*-1,2-dimethylcyclohexane.
- 16) Draw the chair conformer of cyclohexane. Label the axial hydrogens (H<sub>a</sub>) and the equatorial hydrogens (H<sub>e</sub>).
- 17) Draw the most stable conformation of *trans*-1-*tert*-butyl-3-ethylcyclohexane.
- 18) Provide an acceptable name for the compound below.

